

14th ICG SUMMER SCHOOL

GLASS FORMATION, STRUCTURE, AND PROPERTIES



2nd – 7th July 2023 – Montpellier France

Scientific Program



BASIC SCIENCE

	Monday	Tuesday	Wednesday	Thursday	Friday
08h30	Welcome. Introduction to the Course and ICG (J. Parker)				
08h45	Optical absorption and redox chemistry (J. Parker)	Structure (I): Neutron and X-ray diffraction (R. Vacher)	Mechanical properties of glass (I) R. Hand	Modelling (I): atomistic simulations (A. Takada)	Debates
09h45	Thermodynamics I: One-component and multicomponent oxide glasses (R. Conradt)	NMR in oxide glasses (I) (P. Florian)	Glass ceramics (I): (J. Deubener)	Vibrations (I): basics of IR absorption, Brillouin and Raman scattering (B. Hehlen)	Debates
10h45	Coffee break	Coffee break	Coffee break	Coffee break	10h45 Coffee break
11h00	Mass transport in glass. (J. Parker)	Structure (II): Neutron and X-ray diffraction: applications (R.Vacher)	Mechanical properties of glass (II) R. Hand	Modelling (II): Bridging between macroscopic and microscopic phenomena (A. Takada)	Debates
12h00	Thermodynamics II: Combustion processes (R. Conradt)	NMR in oxide glasses (II) (P. Florian)	Glass ceramics (II): (J. Deubener)	Vibrations (II): relation with glass structure (B. Hehlen)	12h Closing Event
13h00	Lunch	Lunch	Lunch	Lunch	
14h30	Students describe their own research activities (5 min /person).	Project allocation & initial preparation	Tutorials (Select from list)	Tutorial (Select from list)	
15h30			Workshops Project preparation	Workshops Project preparation	
18h00	To be announced (E. Muijsenberg)			Chemical durability (R. Conradt)	
19h00	Welcome reception			School Dinner	

TUTORIALS

Glass and phase diagrams - quantitative treatment of multicomponent systems: assessment of glass properties (thermal, mechanical, chemical), approach to structural features & approach to the energetics of glass melting - How to identify the positions of complex glasses in phase diagrams.

Calculating Raman activities : activity of the Raman modes in crystals for a given symmetry and scattering geometry - Molecular selection rules of simple liquids - the case of glasses.

Diffusion: Values of D, examples. Activation energies. Balance of D vs stress relaxation in ion exchange toughening: Optimum temperature range. Significance of $(Dt)^{1/2}$. Examples of time and distance *e.g.* tin bath depth, chemical toughening, chemical durability effects at room T. Crystal growth, nucleation, coarsening.

Practical aspects on atomistic simulations: how to calculate atomic structures and mechanical, transport and optical properties by simulations.

Strength: subject to demand

LIST OF LECTURERS

R. Conradt	Aachen University & uniglassAC GmbH Co.	Aachen, Germany	<i>reinhard.conradt@gmail.com</i>
J. Deubener	Technische Universität Clausthal	Clausthal-Zellerfeld, Germany	<i>jd@tu-clausthal.de</i>
P. Florian	CEMHTI-CNRS	Orleans, France	<i>pierre.florian@cnrs-orleans.fr</i>
R. Hand	University of Sheffield	Sheffield, UK	<i>r.hand@sheffield.ac.uk</i>
B. Hehlen	University of Montpellier	Montpellier, France	<i>bernard.hehlen@umontpellier.fr</i>
E. Muijsenberg	Glass Services	Vsetin, Czech Republic	<i>erik.muijsenberg@gsl.cz</i>
J. Parker	University of Sheffield	Sheffield, UK	<i>j.m.parker@sheffield.ac.uk</i>
A. Takada	University College London (ex-AGC)	Tokyo, Japan	<i>akira_takada_scientist@yahoo.co.jp</i>
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